

TL;DR

Plasticine's super-power is that it can be reshaped in seconds, sticks to things on impact and sinks slowly in syrup. Those three traits unlock at least **seven** physics ideas—from density checks to perfectly inelastic collisions—that frequently appear in IP weighted assessments and H2 practical papers. This article lays out the exact mini-labs, common slip-ups and a 5-day micro-practice plan so you can turn a \$2 block of clay into solid grades.

Exam-Scope Disclaimer

*Stokes' Law is **not** listed as required knowledge in the current IP Physics or H2 Physics syllabuses. Examiners may, however, supply unfamiliar relationships (including Stokes' drag) in an open-ended design or data-handling question and ask you to **derive or apply** them using provided graphs or first-principles ideas. Memorising the Stokes' Law equation is therefore **optional** — keep sub-section 2.2 as an enrichment task, or skip it entirely if you prefer to focus strictly on examinable content.*

1 Why every H2 lab issues a lump of plasticine

- **Malleable & re-usable** - one block can be rolled into spheres, flattened into pucks or packed onto carts in seconds.
- **Safe & classroom-friendly** - no shards, no bounce, no toxic dust.
- **Density** $\approx 1.4 \text{ g} \cdot \text{cm}^{-3}$ —heavy enough to sink in water yet light enough to reach terminal velocity in a 1 m-tall syrup column.
- **Sticky on impact** - guarantees a *perfectly inelastic* collision every time, a requirement in many momentum tasks.

*IP exam setters exploit those virtues because they let students focus on **data handling and uncertainty**, not tricky apparatus.*

2 Seven physics ideas a clay lump can prove

2.1 Density without Archimedes

Roll five different-mass spheres, measure mass (digital balance) and diameter (vernier calliper). A log-log **mass-vs-diameter³ plot** gives the density as the gradient. Students spot linearisation *and* propagate percentage uncertainties in one go.

2.2 Stokes' law in a jam jar

Drop the spheres through glycerine or honey, time the last 10 cm fall. Plot radius² against terminal velocity; gradient yields the fluid viscosity via

$$v_t = \frac{2}{9} \frac{(\rho_{\text{sphere}} - \rho_{\text{fluid}})g}{\eta} r^2.$$

A single kitchen jar turns into an H2-level viscometer.

2.3 Perfectly inelastic collisions on a bench-top

Load a rolling dynamics cart with a motion sensor, fire a plasticine bullet from a spring launcher. The bullet sticks; momentum is conserved but kinetic energy drops —students calculate the lost KE and discuss energy pathways.

2.4 Centre of mass & counter-intuitive balance

Shift tiny lumps along a ruler until it balances on a pencil; plot distance-of-lump vs centre-of-mass position to verify the lever rule. A neat visual for stability questions.

2.5 Elastic vs plastic deformation

Clamp a strip of plasticine and hang masses; strain rises but never returns—an instant demo of plastic region beyond Hooke's law. Compare to a steel spring on the same rig.

2.6 Rolling friction & energy dissipation

Roll an iron ball into a plasticine target on a smooth track; measure the embed depth to estimate the work done by rolling friction. AAPT's experiment shows how multiple concepts intertwine.

2.7 Pressure imprint mapping

Press a loaded test-tube onto a plasticine pad; the contact area reveals pressure distribution—great for qualitative questions on pressure = force/area.

3 Sample WA-style design-question *write – up template*

Prompt

“Design an experiment using plasticine to determine the dynamic viscosity of cooking oil.”

Variable legend

- **Independent variable IV** : sphere radius — the value you deliberately change.
- **Dependent variable DV** : terminal velocity — the outcome you measure.
- **Control variable CV** : oil temperature — kept constant so it does not skew results.

Setup

Section	Key student moves
Diagram	Tall measuring cylinder, stopwatch marks, plasticine sphere
Variables	IV = sphere radius, DV = terminal velocity, CV = oil temperature
Procedure	Release sphere centrally, ignore first 5 cm, time next 20 cm three repeats
Safety & neat tricks	Wear gloves; mark cylinder with masking tape for clear timing gates
Data handling	Plot v vs r^2 ; gradient x constants $\rightarrow \eta$; uncertainty from LINEST

Teachers can mark against H2 practical rubrics: MMO, PDO, ACE.

Rationale & theory

At low Reynolds number ($Re \ll 1$) the net force on a falling sphere equals weight minus buoyancy minus Stokes drag, giving

$$\eta = \frac{2r^2(\rho_{\text{sphere}} - \rho_{\text{oil}})g}{9v_t}$$

Stokes' law is valid only when the flow is laminar and the sphere radius is small compared with the tube diameter.

Equipment list *suggested minima*

- 500 mL graduated measuring cylinder ($\geq 3\text{ cm}$ inner diameter).
- 3 – 5 plasticine spheres, radii 3 mm – 7 mm, measured with a vernier calliper.
- Electronic balance $\pm 0.01\text{ g}$.
- Cooking oil ($\approx 0.065\text{ Pa s}$ at $25\text{ }^\circ\text{C}$).
- Digital thermometer $\pm 0.1\text{ }^\circ\text{C}$ clipped midway down the column.
- Stopwatch capable of 0.01 s or phone camera 120 fps for video timing.

Step-by-step method (justified)

1. **Calibrate & condition:** Warm the oil bath to $25 \pm 0.5\text{ }^\circ\text{C}$ and stir gently to ensure uniform temperature (controls viscosity).
2. **Measure sphere mass & diameter** three times each; compute mean radius and density to $\pm 1\%$.
3. **Release protocol:** Hold the sphere with tweezers at centreline to avoid wall effects, then let go without imparting spin.
4. **Timing window:** Start timer 5 cm below the oil surface (allows acceleration phase to finish) and stop 25 cm lower. Distance is marked with masking tape for consistency.
5. **Repeat** for each radius thrice; discard trials where the sphere touches the wall or creates visible wake (possible $Re > 1$).
6. **Clean-up:** Retrieve spheres with perforated spoon; wash glassware with detergent, dry, store. Follow glassware safety to prevent breakage.

Uncertainty & error analysis

- **Random:** reaction-time error ($\sim 0.2\text{ s}$) divides by long timing distance, so percent uncertainty $\approx 0.2\text{ s} / 4\text{ s} \cong 5\%$. Reduce by video analysis.
- **Systematic:** ignoring buoyancy term overestimates η by $\approx 10\%$. Always subtract ρ_{oil} from ρ_{sphere} .
- **Wall effect:** keep cylinder diameter $\geq 10 \times$ sphere diameter to keep correction $< 1\%$.
- **Temperature drift:** viscosity of vegetable oils changes $\approx 2 - 3\%$; record bath temperature every two trials.

Apply propagation of uncertainty formally (GUM) when reporting η .

Typical data & sample calculation

For a 5.00 mm radius sphere $mass = \rho \times \frac{4}{3}\pi r^3$ falling 0.25 m in 3.60 s at 25 °C:

1. $v_t = 0.25/3.60 = 0.0694 \text{ m} \cdot \text{s}^{-1}$

2. $\rho_{\text{sphere}} = 0.52 / [\frac{4}{3}\pi(2.5 \times 10^{-3})^3] = 1380 \text{ kg} \cdot \text{m}^{-3}$

3. $\eta = \frac{2(2.5 \times 10^{-3})^2(1380 - 920) 9.81}{9 \times 0.0694} = 6.9 \times 10^{-2} \text{ Pa} \cdot \text{s}$

Literature viscosity for canola oil at 25 °C $\approx 0.067 \text{ Pa} \cdot \text{s}$ — within 4 of accepted value.

Validity checks & improvement ideas

- **Check linearity:** plot v against r^2 ; $R^2 > 0.99$ confirms Stokes regime.
- **Lower Re further:** use smaller spheres or colder oil if curvature appears.
- **Automate timing** with Light-Gate + Data-logger to cut human reaction error to < 1
- **Compare liquids:** run the same spheres in water and glycerine to highlight viscosity contrast and reinforce concept transfer.

4 Common mistakes & lightning fixes

Slip-up	Why it hurts	Fix in 10 s
Forgetting buoyancy term in Stokes analysis	Over-estimates viscosity by 10	Write $\rho_{\text{sphere}} - \rho_{\text{fluid}}$ <i>first</i> , highlight in colour
Calling collision “elastic” because carts rebound slightly	Loses theory marks	Stick an extra sliver of clay to guarantee zero rebound
Measuring sphere diameter with a ruler	$\pm 1 \text{ mm}$ error dominates	Use a vernier; take three perpendicular readings
Rounding mid-calculation	Data scatter inflates	Keep one extra s.f. until final line

5 5-Day micro-practice sprint

Day	15-min mission	Concept locked
1	Roll 3 spheres, plot mass vs diameter ³	Density & linearisation
2	Drop one sphere in honey, record video at 120 fps	Terminal velocity
3	Glue clay to air-track glider, do a sticky collision	Momentum conservation
4	Balance a ruler with clay lumps, sketch CoM shift	Centre of gravity
5	Quiz yourself: list every uncertainty source seen	PDO reflex

Tick each box, snap a photo, post to class chat—peer accountability matters.

6 Quick FAQ

Q Why does IP love “plasticine questions”?

Because the same blob lets exam writers weave density, kinematics and mechanics into one neat package, mirroring the cross-topic flavour of A-Level practicals.

Q Won't the clay absorb oil and change mass?

Mass change over a 30 s run is < 0.1

— well inside typical measurement uncertainty.

Q Is Blu-Tack a valid substitute?

Blu-Tack is visco-elastic; rebound spoils perfectly inelastic assumptions. Stick to plasticine.

7 Further reading & ready-to-borrow ideas

1. [Optional Practical - Density of Plasticine | InThinking IB Physics](#)
2. [Stokes' law | Wikipedia](#)
3. [Measuring Rolling Friction with Plasticine | The Physics Teacher \(AIP\)](#)
4. [H2 Physics Syllabus 9478 | SEAB](#)
5. [H2 Physics Practical Discussion | Reddit r/SGExams](#)
6. [Density of Plasticine Lab Report | Scribd](#)
7. [Hooke's Law and Plastic Materials | Physics Forums](#)
8. [Air-Track Inelastic Collisions Demo 24.12 | UCSB Physics](#)
9. [Momentum Conservation: Shooting Clay Balls at a Cart | YouTube - Zak's Lab](#)
10. [Centre of Gravity Balance Trick | YouTube - Science Experiments](#)

Next step: grab 50 g of plasticine, run **one** experiment tonight, and log your biggest surprise.